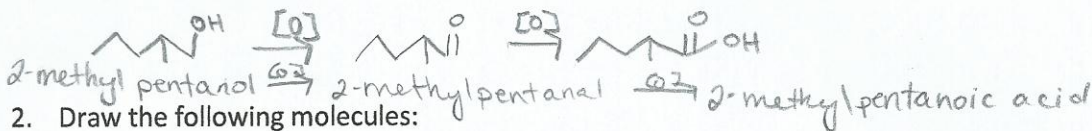
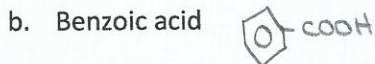
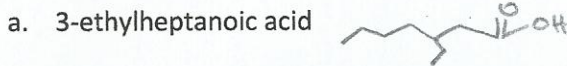


1. What are the reactants that form 2-methylpentanoic acid? Draw this oxidation reaction.



2. Draw the following molecules:



3. Describe the overall solubility as well as the solubility trends of carboxylic acids.

they are very soluble; solubility decreases as the alkyl chain increases (gets longer past 4 carbon chain)

4. Number the following molecules in terms of highest (1) to lowest (5) in terms of boiling point.

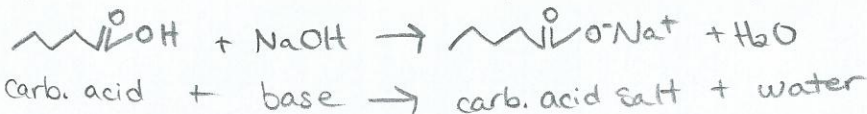


5. For number 4, tell which ones are water soluble and which ones are less water soluble.

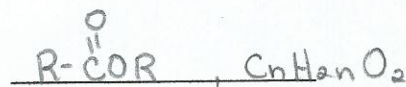
-not soluble: pentane - water soluble - none

-slightly: pentanoic acid, benzoic acid, pentanal, pentanol

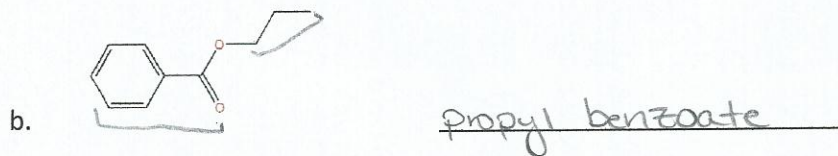
6. Show the reaction that forms the carboxylic acid salt **sodium pentanoate**.



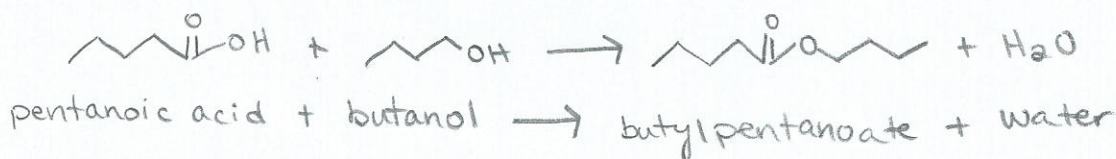
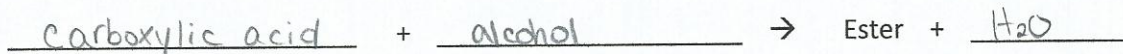
7. Write the R formula and molecular formula for an ester.



8. Name the following esters:

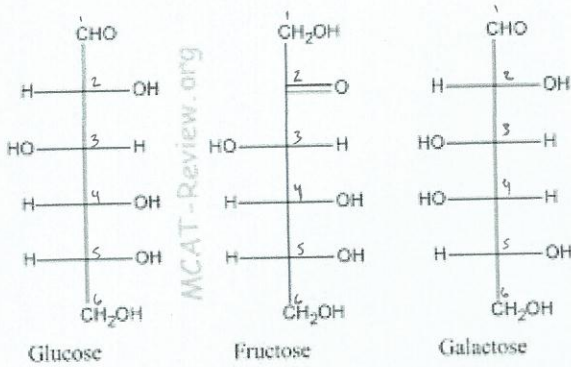


9. Complete the following phrase. Then complete the reaction that will create butyl pentanoate.

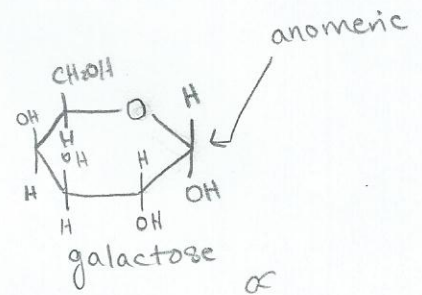
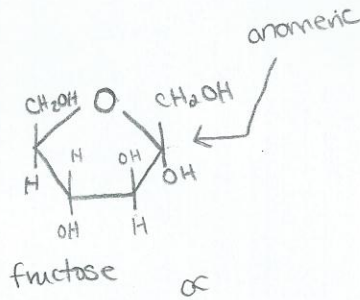
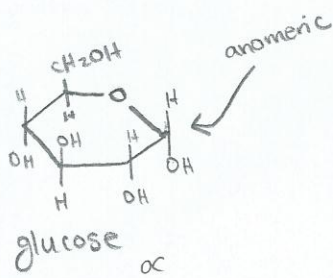


b.p. carbox. acid > alcohol > ald. / ket. > ether, alkane

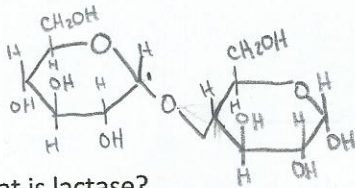
Review of Carbohydrates- MWF only



1. Put glucose, fructose, and galactose into individual rings. Identify the anomeric carbons, and tell whether it is alpha or beta.



2. Link two glucose molecules together to form a disaccharide. Make it have an α -1,4 linkage.



3. What is lactase?

an enzyme that breaks the β -1,4 linkage in lactose. If it is not present to break lactose, lactose is fermented in the large intestine, causing bloating, cramps etc.

4. Are lactose free products sweeter or less sweet than those with lactose?

sweeter because they have galactose + glucose present in them.